# rayny

# Highly Regio- and Stereoselective Synthesis of Boron-Substituted Enynes via Copper-Catalyzed Borylation of Conjugated Diynes

DingXi Li, Yeong Eun Kim, and Jaesook Yun\*

Department of Chemistry and Institute of Basic Science, [Su](#page-2-0)ngkyunkwan University, Suwon 440-746, Korea

**S** Supporting Information

[AB](#page-2-0)STRACT: [A mild copp](#page-2-0)er-catalyzed regio- and stereoselective monoborylation of conjugated diynes with bis- (pinacolato)diboron that affords enynylboronates is reported. The reaction is efficient for different types of conjugated diynes including unsymmetrical diynes and produces enynylboron compounds with high and complementary regioselec-



tivity compared with classical hydrometalation reactions. In particular, the reactions of internal conjugated diynes with a silyl substitution produced highly functionalized enynes with high regio- and stereoselectivity, which can be used in further transformations.

O rganoboron compounds are versatile reagents, and vinylboronates represent especially useful reagents in synthetic organic chemistry, as evidenced by their use in transition-metal catalyzed cross-coupling reactions.<sup>1</sup> Hydroboration of alkynes is one of the most convenient and straightforward methods of producing a variety [o](#page-2-0)f vinylboronates. $2,3$  However, control of regio- and stereochemistry is a frequent obstacle faced in hydroboration.

Organi[c m](#page-2-0)olecules with extended  $\pi$ -conjugation have garnered much attention due to their wide application as important building blocks for designing nanostructures, polymer synthesis, and organic synthesis.<sup>4</sup> In this context, the development of efficient methods for the synthesis of enynylboronates with high regio- and [st](#page-2-0)ereoselectivity is of considerable importance. While the hydroboration of monoynes has been widely studied, reports on the hydroboration of conjugated diynes are scarce.<sup>5</sup> Indeed, there has been only one study by Polston et al. on the selective hydroboration of conjugated diynes with di[al](#page-3-0)kylboranes as a hydroborating reagent.<sup>3a</sup> In the report, the monohydroboration of symmetrically alkyl-substituted diynes with disiamylborane afforded enynes [wi](#page-3-0)th boron in the internal positions of the diyne system which are less sterically hindered positions (Scheme 1). Other related hydrometalation processes such as hydrosilylation<sup>6</sup> and





hydrostannation $<sup>7</sup>$  offered similar regioselectivities with the</sup> metallic moiety (E) in internal positions. Therefore, development of synt[he](#page-3-0)tic methods displaying complementary regioselectivity to existing methods is valuable.

We have previously shown that catalytic borylcupration (Cu−B) to C−C multiple bonds and protonolysis by methanol can produce hydroborylated products.<sup>8</sup> In contrast to coppercatalyzed borylation of monoynes,<sup>9</sup> the selective reaction of [d](#page-3-0)iynes has not yet been reported.<sup>10</sup> Indeed, selective monoborylation of diynes is quit[e](#page-3-0) challenging because the borylenyne products still possess reacti[ve](#page-3-0) π-conjugated C−C multiple bonds in the presence of the nucleophilic Cu−B catalyst, in contrast to other  $\pi$ -conjugated starting substrates.<sup>10</sup> Herein, we present our results on the monoborylation of conjugated diynes to produce enynylboron compounds wi[th](#page-3-0) the boron at the end position of various diynes.

We initially investigated the reaction of 1,4-diphenylbutadiyne (1a) as a model substrate under various reaction conditions (Table 1). The reaction was optimized by the combination of a catalytic copper chloride, NaOt-Bu, and  $P(p-$ tol)<sub>3</sub> in THF at 11 °[C](#page-1-0). The reaction yielded a high conversion in 3 h and afforded the monoborylated product 2a as the major product (entry  $1$ ).<sup>11</sup> The structure of  $2a$  was confirmed by derivatizing it to the known structure, $12$  and the regioselectivity was found to be [com](#page-3-0)plementary to the conventional hydroboration reaction, $5$  with the boron in [e](#page-3-0)xternal positions of the diyne system. Reactions using the weaker electron-donating  $P(OEt)$ <sub>3</sub> or PPh<sub>3</sub> [li](#page-3-0)gands resulted in a decreased yield or longer reaction time (entries 2 and 3). Use of a bidentate ligand or other bases was not beneficial to the reaction (entries 4 and 5). Controlling the amount or concentration of the diboron reagent or the temperature was not effective in increasing the efficiency of the reaction (entries 6−8).

Received: December 26, 2014 Published: February 9, 2015



<span id="page-1-0"></span>Table 1. Screening under Various Conditions



a Determined by GC analysis with an internal standard based on consumption of 1a. <sup>b</sup>NMR yield was determined using DMF as an internal standard. "Isolated yield was obtained by column chromatography at low temperature  $(0 °C)$ .  $\alpha$ Xantphos = 9,9<sup>7</sup>-dimethyl-4,5bis(diphenylphosphino)xanthene.

We next investigated the reaction of symmetrical diynes<sup>13</sup> with various aryl or alkyl substituents (Table 2). Gratifyingly, the reaction with the diynes afforded the desired monobo[ry](#page-3-0)lated enyne products without the other regioisomer.<sup>14</sup> The Bsubstituted enyne products were isolated in good yields as the Bpin derivatives  $(2)$  or the corresponding and m[ore](#page-3-0) stable trifluoroborate salts (3). Diyne substrates with electronwithdrawing or -donating aryl substituents were suitable substrates for the reaction (entries 2 and 3). The more sterically hindered o-tolyl-substituted diyne (1d) also afforded the product in good yield (entry 4). The reaction of electronrich diynes with alkyl or electron-donating aryl substituents proceeded well with  $P(OEt)$ <sub>3</sub> as the ligand at room temperature. Primary (1e) and secondary (1f) alkyl-substituted diynes reacted smoothly to give the desired products with the boron in the external position, but the tertiary alkyl-substituted diyne (1g) afforded the product 3g with the opposite regioselectivity (entries  $5-\hat{8}$ ).<sup>15</sup> The results indicated that severe steric bulkiness of the tert-butyl group of diynes could alter the regioselectivity of the [cu](#page-3-0)rrent catalytic reaction.

We further investigated the monoborylation of unsymmetrically substituted conjugated diynes, the results of which are shown in Table 3. Surprisingly, the reaction of 1-phenyl-1,3 pentadiyne (4a) afforded 5a with excellent regioselectivity with the boron at the [e](#page-2-0)xternal position close to the methyl group (Table 3, entry 1). The reaction of primary alkyl-substituted diyne 4b gave product 5b with the same type of regioselectivity as that [o](#page-2-0)f 4a (entry 2). In contrast to the aryl-substituted conjugated diynes (4a and 4b), silyl-substituted diynes (4c and 4d) showed an alternative regioselectivity in which the boron was attached at the external carbon positioned away from the silyl group via regioselective borylcupration to the more electrophilic triple bond (entries  $3-5$ ).<sup>16</sup> These results indicate that the silyl group acted as a strong directing group. In all of these examples, the boron addition [occ](#page-3-0)urred at the external positions with excellent regioselectivity with no observation of the other regioisomers. Finally, other diyne substrates such as terminal or electronically different aryl-substituted diynes were investigated. The terminal diyne 4e gave the monoborylation product in good yield without significant formation of the



Table 2. Regioselective Monoborylation of Symmetrical

<sup>a</sup>The reaction was conducted with the combination of 5 mol % CuCl, 6 mol %  $P(p$ -tol)<sub>3</sub>, 10 mol % NaOt-Bu, and 2 equiv of MeOH in THF at 11 °C for 2–4 h. <sup>b</sup>Isolated yield of 2 or 3. See Supporting Information for details  ${}^{\circ}$ PPh<sub>3</sub> was used.  ${}^{\circ}P$ (OEt)<sub>3</sub> ligand was used at room tem[perature.](#page-2-0)  ${}^{e}$ Reaction was performed at room temperature.

diborylated compound (entry 6). However, diyne 4f with different aryl substituents resulted in poor regioselectivity and produced both regioisomeric products in a 1:1 ratio<sup>17</sup> (entry 7), which indicated no strong directing bias between the two aryl groups.

Synthetic applications of boron-substituted enynes were carried out to illustrate their usefulness in organic synthesis (Scheme 2). Palladium-catalyzed cross-coupling of borylenynes (2a and 5a) with iodobenzene gave the corresponding enynes12,18 in good yields. Protonolysis of 3a gave an 86% yield of t[he](#page-2-0) corresponding cis-enyne 9. Selective desilylation of 6c was [poss](#page-3-0)ible to yield 10 with a terminal alkynyl group, which

#### <span id="page-2-0"></span>Table 3. Monoborylation of Unsymmetrical Conjugated  $Divnes^a$



<sup>a</sup>The reaction was conducted with the combination of 5 mol % CuCl, 6 mol %  $P(OEt)_{3}$ , and 10 mol % NaOt-Bu in THF at 11 °C or room temperature for  $2-6$  h.  $\frac{b}{2}$  Isolated yield of 5 or 6. <sup>c</sup>Isolated yield of a mixture of 6f and 6f′.



showed a complementary regioselectivity to that of 5e, obtained from the monoborylation of the terminal diyne (Table 3, entry 6).

In summary, we have developed a copper-catalyzed regioand stereoselective monoborylation of conjugated diyne compounds that affords boron-containing enynes. The reaction is efficient for different types of conjugated diynes including unsymmetrical diynes. The regioselectivity obtained was complementary to that of classical hydrometalation reactions, with the boron moiety at the external carbons of conjugated diynes. In particular, the reactions of silyl-substituted conjugated diynes produced highly functionalized enynes with high regio- and stereoselectivity, which can be used in further transformations.

# ■ ASSOCIATED CONTENT

#### **6** Supporting Information

Experimental procedures and characterization of products and copies of the  ${}^{1}H$  and  ${}^{13}C$  NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

### ■ AUTHOR INFORMATION

#### Corresponding Author

\*E-mail: jaesook@skku.edu.

# Notes

The authors declare no competing financial interest.

#### ■ ACKNOWLEDGMENTS

This research was supported by the Basic Science Research Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Education, Science, and Technology (NRF-2013R1A1A2058160).

#### ■ REFERENCES

(1) (a) Contemporary Boron Chemistry; Davidson, M. G., Hughes, A. K., Marder, T. B., Wade, K., Eds.; Royal Society of Chemistry: Cambridge, 2000. (b) Organoboranes for Syntheses; ACS Symposium Series 783; Ramachandran, P. V., Brown, H. C., Eds.; American Chemical Society: Washington, DC, 2001. (c) Boronic Acids: Preparation and Applications in Organic Synthesis, Medicine and Materials, 2nd ed.; Hall, D. G., Ed.; Wiley-VCH: Weinheim, 2011. (d) Mkhalid, I. A. I.; Barnard, J. H.; Marder, T. B.; Murphy, J. M.; Hartwig, J. F. Chem. Rev. 2010, 110, 890−931.

(2) (a) Pereira, S.; Srebnik, M. Organometallics 1995, 14, 3127−3128. (b) Ohmura, T.; Yamamoto, Y.; Miyaura, N. J. Am. Chem. Soc. 2000, 122, 4990−4991. (c) Gunanathan, C.; Hölscher, M.; Pan, F.; Leitner, W. J. Am. Chem. Soc. 2012, 134, 14349−14352. (d) Gridnev, I. D.; Miyaura, N.; Suzuki, A. Organometallics 1993, 12, 589−592. (e) Barbeyron, R.; Benedetti, E.; Cossy, J.; Vasseur, J.-J.; Arseniyadis, S.; Smietana, M. Tetrahedron 2014, 70, 8431−8452.

(3) Other examples of vinylboronate synthesis: (a) Ohmura, T.; Suginome, M. Bull. Chem. Soc. Jpn. 2009, 82, 29−49. (b) Al Quntar, A. A. A.; Botvinik, A.; Rubinstein, A.; Srebnik, M. Chem. Commun. 2008, 5589−5591. (c) Mannathan, S.; Jeganmohan, M.; Cheng, C.-H. Angew. Chem., Int. Ed. 2009, 48, 2192−2195. (d) Takagi, J.; Takahashi, K.; Ishiyama, T.; Miyaura, N. J. Am. Chem. Soc. 2002, 124, 8001−8006. (e) Jiao, J.; Hyodo, K.; Hu, H.; Nakajima, K.; Nishihara, Y. J. Org. Chem. 2014, 79, 285−295. (f) Takaya, J.; Kirai, N.; Iwasawa, N. J. Am. Chem. Soc. 2011, 133, 12980−12983. (g) Mkhalid, I. A. I.; Coapes, R. B.; Edes, S. N.; Coventry, D. N.; Souza, F. E. S.; Thomas, R. L.; Hall, J. J.; Bi, S.-W.; Lin, Z.; Marder, T. B. Dalton Trans. 2008, 1055−1064. (4) (a) Roncali, J. Chem. Rev. 1997, 97, 173−205. (b) Nielsen, M. B.; Diederich, F. Chem. Rev. 2005, 105, 1837−1867. (c) Galm, U.; Hager,

<span id="page-3-0"></span>(5) (a) Zweifel, G.; Polston, N. L. J. Am. Chem. Soc. 1970, 92, 4068− 4071. (b) For catalytic diborations of conjugated diynes: (b) Lesley, G.; Nguyen, P.; Taylor, N. J.; Marder, T. B.; Scott, A. J.; Clegg, W.; Norman, N. C. Organometallics 1996, 15, 5137−5154. (c) Bauer, F.; Braunschewig, H.; Gruss, K.; Kupfer, T. Organometallics 2011, 30, 2869−2884.

(6) Alonso, F.; Buitrago, R.; Moglie, Y.; Sepúlveda-Escribano, A.; Yus, M. Organometallics 2012, 31, 2336−2342.

(7) Zhang, H. X.; Guibé, F.; Balavoine, G. J. Org. Chem. 1990, 55, 1857−1867.

(8) (a) Mun, S.; Lee, J.-E.; Yun, J. Org. Lett. 2006, 8, 4887−4889. (b) Lee, J.-E.; Kwon, J.; Yun, J. Chem. Commun. 2008, 733−734. (c) Kim, H. R.; Jung, I. G.; Yoo, K.; Jang, K.; Lee, E. S.; Yun, J.; Son, S. U. Chem. Commun. 2010, 46, 758−760. (d) Jung, H.-Y.; Yun, J. Org. Lett. 2012, 14, 2606−2609.

(9) For reviews, see: (a) Yun, J. Asian J. Org. Chem. 2013, 2, 1016− 1025. (b) Fujihara, T.; Semba, K.; Terao, J.; Tsuji, Y. Catal. Sci. Technol. 2014, 4, 1699−1709. (c) Alfaro, R.; Parra, A.; Alemán, J.; Tortosa, M. Synlett 2013, 24, 804−812. (d) Stavber, G.; Č asar, Z. ChemCatChem 2014, 6, 2162−2174.

(10) For copper−boron catalysis of enynes, see: (a) Sasaki, Y.; Horita, Y.; Zhong, C.; Sawamura, M.; Ito, H. Angew. Chem., Int. Ed. 2011, 50, 2778-2782. For dienes: (b) Sasaki, Y.; Zhong, C.; For dienes: (b) Sasaki, Y.; Zhong, C.; Sawamura, M.; Ito, H. J. Am. Chem. Soc. 2010, 132, 1226−1227. For allenes: (c) Yuan, W.; Ma, S. Adv. Synth. Catal. 2012, 354, 1867−1872. (d) Semba, K.; Shinomiya, M.; Fujihara, T.; Terao, J.; Tsuji, Y. Chem.Eur. J. 2013, 19, 7125−7132.

(11) Deboronated and diborylated products (<10%) were observed. (12) Compound 7 and related regioisomers: (a) Fu, X.; Liu, Y.; Li, Y. Organometallics 2010, 29, 3012−3018. (b) Chary, B. C.; Kim, S.; Shin, D.; Lee, P. H. Chem. Commun. 2011, 47, 7851–7853. (c) Liu, Y.; Zhong, Z.; Nakajima, K.; Takahashi, T. J. Org. Chem. 2002, 67, 7451− 7456. (d) Djakovitch, L.; Rollet, P. Adv. Synth. Catal. 2004, 346, 1782−1792.

(13) (a) Zhang, S.; Liu, X.; Wang, T. Adv. Synth. Catal. 2011, 353, 1463−1466. (b) Nishihara, Y.; Ikegashira, K.; Hirabayashi, K.; Ando, J.-i.; Mori, A.; Hiyama, T. J. Org. Chem. 2000, 65, 1780−1787.

(14) A crude reaction mixture was analyzed by NMR spectroscopy for detection of regioisomers.

(15) The structures of 2e and 2f were confirmed by NMR coupling constants of the vinylic protons demonstrating allylic coupling as opposed to vicinal coupling. The structure of 3g was confirmed by comparing its vinylic proton with that of the corresponding trifluoroborate salt obtained by hydroboration as in ref 5 and  $KHF_2$ treatment.

(16) Zweifel, G.; Leong, W. J. Am. Chem. Soc. 1987, 109, 6409−6412. (17) The ratio  $(1:1)$  was determined by <sup>1</sup>H NMR analysis of a crude reaction mixture of 5f and 5f′. The crude mixture was treated with  $KHF<sub>2</sub>$ , and isolation of 6f and 6f' afforded a regioisomeric mixture of

(18) Wen, Y.; Wang, A.; Jiang, H.; Zhu, S.; Huang, L. Tetrahedron Lett. 2011, 52, 5736−5739.

67:33 in 72% yield.